


EELE408 Photovoltaics

Lecture 06: Semiconductors Advanced Topics

Dr. Todd J. Kaiser
 tjkaiser@ece.montana.edu

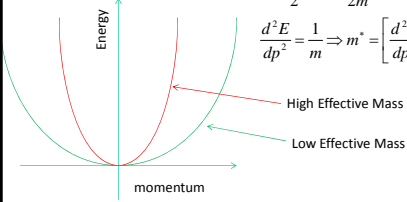
Department of Electrical and Computer Engineering
 Montana State University - Bozeman



Energy of electrons in free space

- Continuum of energy states – electrons can have any energy on parabola
- Mass of electron relates to the curvature or second derivative of the energy-momentum plot

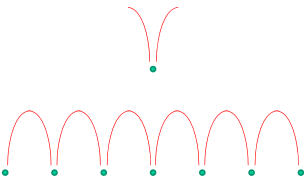
$$E = \frac{1}{2}mv^2 = \frac{p^2}{2m} \quad (p = mv)$$

$$\frac{d^2 E}{dp^2} = \frac{1}{m} \Rightarrow m^* = \left[\frac{d^2 E}{dp^2} \right]^{-1} \quad \text{Effective Mass}$$


2

Periodic Potentials

$$V(r) = \frac{-q^2}{4\pi\epsilon_0 r}$$

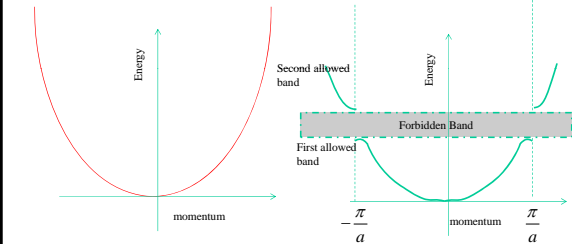
$$V(r) = V(r \pm na)$$


“When I started to think about it, I felt that the main problem was to explain how the electrons could sneak by all the ions in a metal.... By straight Fourier analysis I found to my delight that the wave differed from the plane wave of free electrons only by a periodic modulation” F. Bloch

3

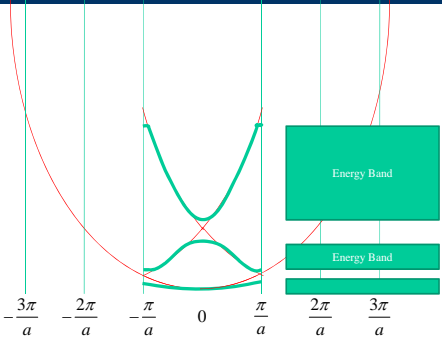
Forbidden Band

- The periodic potential requires that the electrons can not be at the atom positions. This creates forbidden energy levels.



4

Dispersion Curves

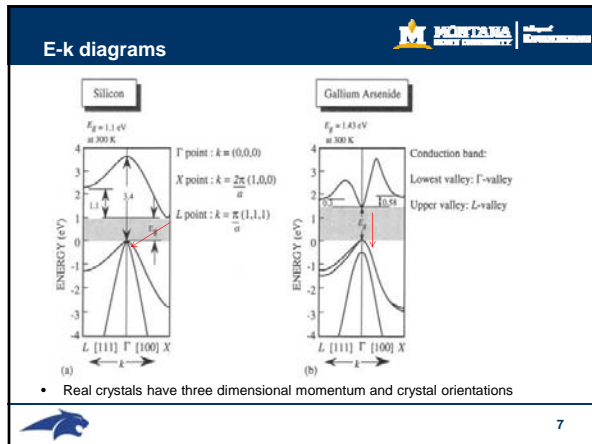


5

Schrodinger Equation for Crystals

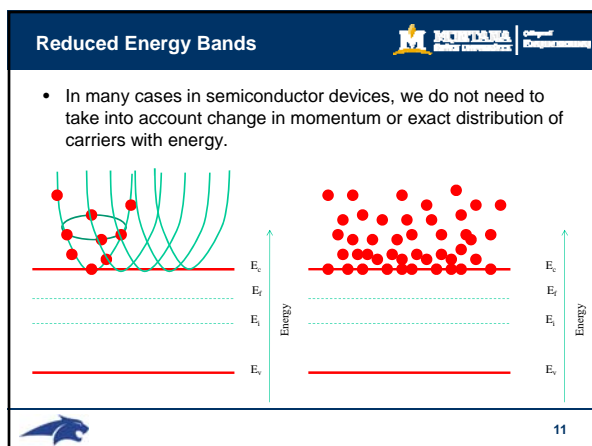
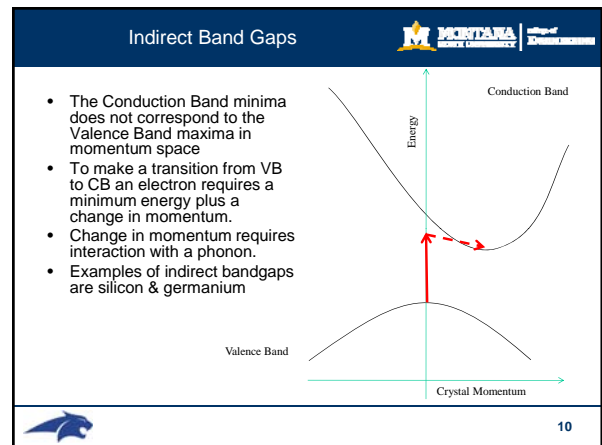
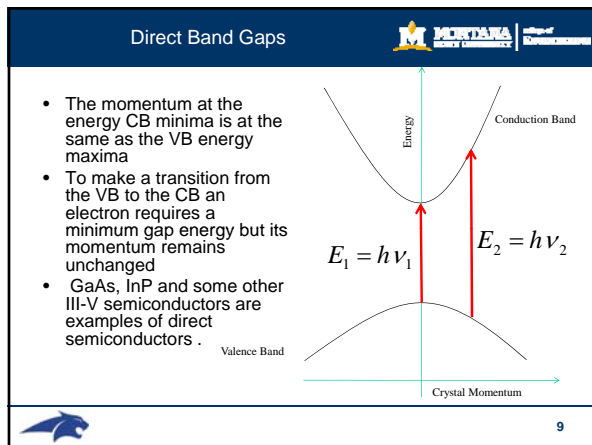
- Solution shows some energies have multiple energy states while some energies are disallowed.
 - Electrons can only occupy places “on” the E-k diagram
 - The band gap is minimum difference between the two uppermost bands (conduction and valence band)
- Near the minima or maxima, bands are approximately parabolic, and hence can be approximated as “free” electrons
- Effective mass of each band near zone center varies, depending on its curvature
- Effective mass varies with k.

6



Consequences of E-k

- The alignment of the conduction band minimum and valence band maxima have a major impact on electrical properties such as absorption and recombination
- Aligned maxima and minima are called **direct band gaps**, misaligned are called **indirect band gaps**.



Probability of State Occupation

- Pauli Exclusion Principle → each allowed energy level can only hold 2 electrons (Spin Up and Spin Down)
- At low temperatures all available states are filled up to a certain energy level called the Fermi level (E_F)
- As temperature rises some electrons gain energy above the Fermi level.
- The probability of occupation is statistically given by the Fermi-Dirac Distribution Function

Fermi-Dirac Distribution

$$f(E,T) = \frac{1}{1 + e^{\frac{E-E_F}{kT}}}$$

- The fermi level is the energy where the probability of the state being filled is at 50%
- 3kT above fermi level probability effectively zero
- 3kT below the fermi level probability effectively one

$T=0 < T_1 < T_2$

Electronic Materials

- Metals
 - Fermi level lies within an allowed band
- Insulators
 - One band fully filled (cannot conduct the charge is full)
 - Large band gap (few conduction electrons)
 - Fermi in the band gap
- Semiconductor
 - Insulator with a small band gap
 - Fermi-Dirac distribution spans gap at elevated temperatures

Material Classification

Insulator Metal Semiconductors Semimetal

Effective Mass values for common semiconductors

		Silicon	Germanium	Gallium Arsenide
Band gap (300K)	E_g (eV)	1.12	0.66	1.424
Electron Effective Mass	m_e^*/m_0	1.08	0.55	0.066
Hole Effective Mass	m_h^*/m_0	0.81	0.36	0.52

Densities of Electrons and Holes

- Low Temperature: no electrons in conduction band (not a conductor)
- Medium Temperature: some electrons in conduction band (poor conductor)
- High Temperature: more electrons in conduction band (better conductor)

Electron and Hole Concentrations

$$n = N_C e^{\frac{E_F - E_C}{kT}}$$

$$p = N_V e^{\frac{E_V - E_F}{kT}}$$

- N_C : effective density of states in the conduction band $\sim 10^{19}/\text{cm}^3$
- N_V : effective density of states in the valence band $\sim 10^{19}/\text{cm}^3$
- Function of material

